

Estd. June 1964



“Education for Knowledge, Science and Culture.”

– Shikshanmaharshi Dr. Bapuji Salunkhe

Shri. Swami Vivekanand Shikshan Sanstha's

**VIVEKANAND COLLEGE, KOLHAPUR
(AUTONOMOUS)**

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UGC Recognition Under 2 F & 12(B) UGC Act 1956

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Department of Chemistry


Course Outcomes (COs): Chemistry

B. Sc. Part III Chemistry (Introduced in the year 2020-21)	
Semester V	
Paper V: DSC-1002 E1	
Section I: Physical Chemistry	
CO No.	On completion of the course, student will be able to:
CO1	Understand the wave mechanics of atomic structure.
CO2	Know the phenomenon related to the micro particle like electrons.
CO3	Acquire the fundamentals behinds the spectroscopic techniques like Raman, electronics and vibrational spectroscopy.
CO4	Adopt the basics of photochemistry, structure, defects in crystals and theory of the reaction rates.
Section II: Inorganic Chemistry	
CO1	Describe the synthesis and applications of the semiconductors and superconductors in electrical and electronic devices.
CO2	Impart essential knowledge regarding classification, types, mechanism and applications of catalyst in industrial fields.
CO3	Improve the level of understanding of structure, method of preparation and applications of organometallic compounds in various fields.
CO4	Gain thorough knowledge of role of various metals and nonmetals in our health.
Paper VI: DSC-1002 E2	
Section I: Organic Chemistry	
CO No.	On completion of the course, student will be able to:
CO1	Learn mechanism of different organic name reactions and to

	become confident to solve the problems based on the reactions.
CO2	Adopt the utility of reagents in organic synthesis.
CO3	Understand the fundamentals of terpenoids, alkaloids and applications of nucleophilic substitution reactions of aromatic compounds.
CO4	Acquire the knowledge of pharmaceuticals and its use.
Section II: Analytical Chemistry	
CO1	Discuss the basic concepts of qualitative and quantitative analysis.
CO2	Develop the skills of potentiometric, titrimetric, gravimetric and colorimetric analysis.
CO3	Illustrate the separation technique paper chromatography.
CO4	Understand the basic of thin layer chromatography.
Semester VI	
Paper VII: DSC-1002 F1	
Section I: Physical Chemistry	
CO No.	On completion of the course, student will be able to:
CO1	Understand the theoretical aspect of chemical transformation.
CO2	Recognize about surface phenomenon and isotherms of surface reactions.
CO3	Describe the methods of detections of radioactivity of the samples, the working principle of cells and batteries.
CO4	Explain the chemistry behind the ethanol fermentation by anaerobic bacteria.
Section II: Inorganic Chemistry	
CO1	Get idea about theories, factors and knowledge of prevention from corrosion.
CO2	Gain the knowledge about ligands, chelates, classification and applications of chelating agents in analytical chemistry.
CO3	Develop interest in various nuclear reactions and role of radio isotopes in medicinal, industrial and archaeology fields.
CO4	Study the important aspects of the mechanism of the reactions involved in inorganic complexes of transition metals. Also acquire a basic understanding of nanochemistry, nanotechnology and its fascinating aspects.
Paper VIII: DSC-1002 F2	
Section II: Organic Spectroscopic Techniques	

CO No.	On completion of the course, student will be able to:
CO1	Understand the basic concepts of spectroscopy.
CO2	Acquire the knowledge of spectroscopic techniques such as NMR and Mass Spectroscopy.
CO3	Interpret molecular structures by using spectroscopic techniques.
CO4	Adopt the knowledge of various spectroscopic techniques such as UV and IR.
Section II: Industrial Chemistry	
CO1	Understand the basics of industrial chemistry.
CO2	Adopt the knowledge of sugar and jaggery industry.
CO3	Learn and understand processes involved in manufacturing of alcohol and heavy chemicals.
CO4	Illustrate the overall information regarding manufacture of fertilizers.




 Dr. (Mrs). S, D, Shirke
HEAD
 DEPARTMENT OF CHEMISTRY
 VIVEKANAND COLLEGE, KOLHAPUR
 (EMPOWERED & AUTONOMOUS)