

“Dissemination of Education through Knowledge, Science and Culture”

-Shikshanmaharshi Dr. Bapuji Salunkhe

Shri Swami Vivekanand Shikshan Sanstha, Kolhapur's

## **Vivekanand College, Kolhapur**

**(Empowered Autonomous)**

(Affiliated with Shivaji University, Kolhapur)

### **DEPARTMENT OF CHEMISTRY**

#### **TEACHING PLAN**

**(2023-24)**

<b>Sr. No.</b>	<b>Name of Faculty</b>	<b>Designation</b>
1	Dr. Mrs. S. D. Shirke	Head of Department
2	Prof. Dr. A. S. Kumbhar	Professor
3	Mr. S. S. Kadam	Assistant Professor
4	Dr. S. S. Ankushrao	Assistant Professor
5	Dr. Mrs. S. D. Shinde	Assistant Professor
6	Dr. Mrs. A. S. Tapase	Assistant Professor
7	Dr. D. S. Gaikwad	Assistant Professor
8	Dr. A. A. Patravale	Assistant Professor
9	Mr. A. T. Mane	Assistant Professor
10	Dr. Ms. S. D. Kharade	Assistant Professor
11	Dr. Ms. A. S. Rajmane	Assistant Professor
12	Ms. S. N. Inamdar	Assistant Professor
13	Ms. P. A. Gholap	Assistant Professor
14	Ms. V. S. Rajmane	Assistant Professor



# Vivekanand College, Kolhapur (Empowered Autonomous)

## Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. II, Sem- IV

Department: Chemistry

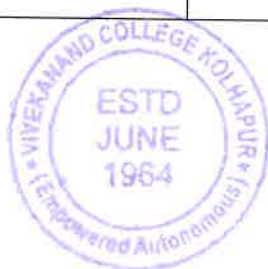
Subject: Chemistry

Course Title: DSE-1002E2 : Organic & Analytical Chemistry

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	1. Amines and Diazonium Salts	<ul style="list-style-type: none"> <li>Theory :Introduction</li> <li>Amines- Definition</li> <li>Classification</li> <li><b>Practicals</b> - Conductometry -02, organic spotting and organic preparation</li> </ul>
01	08	09		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Continued...	<ul style="list-style-type: none"> <li>Theory -Preparation of Primary , secondary and Tertiary amines.- Chemical Reactions</li> <li><b>Practicals</b>- Chemical Kinetics, and Viscosity</li> <li>Preparation of Mohr's salt</li> <li>Semi-micro mixture analysis- Two</li> </ul>
01	32	33		
Month: February			Module/Unit:	Sub-units planned
02	32	34	Continued...	<ul style="list-style-type: none"> <li>Continued...Diazonium Salts-Definition, exxamples</li> <li>Preparation of diazonium salts from aromatic amines and HNO<sub>2</sub></li> <li>Electrophilic Substitution reactions and Sandmeyer's reactions.</li> <li><b>Practicals</b> – Semimicro analysis- 1 mixture</li> </ul>
Month: March			Module/Unit:	Sub-units planned
-	-	-	-	Reaction of diazonium salt with beta naphthol, and different electrophiles.
				Semimicro Mixture Analysis : 02 mixtures
	Month: April			Theory and Practical Examinations.

*S.D. Shirke*

Dr. S. D. Shirke



*S.D. Shirke*

Dr. S. D. Shirke

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# Vivekanand College, Kolhapur (Empowered Autonomous)

## Annual Teaching Plan

Academic Year: 2023-24 [Semesters: B.Sc. I (Major/Minor), Sem-I] Department: Chemistry

Subject: Chemistry Course Title: DSC03CHE11/MIN03CHE11: Inorganic Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: July			Module/Unit:	Sub-units planned
Lectures	Practicals hr	Total	Induction	<ul style="list-style-type: none"> <li>• General Introduction</li> <li>• Discussion on Course Structure</li> <li>• Discussion on Examination pattern</li> <li>• Discussion on Syllabus</li> </ul>
06	-	06		
Month: August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemical Bonding and Molecular structure -Ionic Bonding	<ul style="list-style-type: none"> <li>• General introduction</li> <li>• Types of Bond</li> <li>• Formation of ionic Solid</li> <li>• Factors Governing to Formation of ionic Solid</li> </ul>
12	16	28		
Month: September			Module/Unit:	Sub-units planned
12	16	28	Chemical Bonding and Molecular structure -Ionic Bonding	<ul style="list-style-type: none"> <li>• Born-Haber Cycle</li> <li>• Applications of Born-Haber Cycle</li> <li>• Fajan's Rule</li> <li>• Applications of Fajan's rule</li> <li>• % of Covalent Character in Ionic Comp.</li> </ul>
Month: October			Module/Unit:	Sub-units planned
12	16	28	Chemical Bonding and Molecular structure-Valence bond theory (VBT)	<ul style="list-style-type: none"> <li>• Valence Bond Theory: Introduction, Assumptions, Applications and Limitations.</li> <li>• Concept of hybridization, different types of hybridization and geometry of molecule.</li> <li>• Linear geometry <math>\text{BeCl}_2</math> (<math>sp</math> hybridization)</li> <li>• Planer trigonal geometry <math>\text{BF}_3</math> (<math>sp^2</math> hybridization)</li> <li>• Tetrahedral geometry <math>\text{SiCl}_4</math> (<math>sp^3</math> hybridization)</li> </ul>
Month: November			Module/Unit:	Sub-units planned
06	08	14	Chemical Bonding and Molecular structure-Valence bond theory (VBT)	<ul style="list-style-type: none"> <li>• Trigonal bipyramidal geometry <math>\text{PCl}_5</math> (<math>sp^3d</math> hybridization)</li> <li>• Octahedral geometry <math>\text{SF}_6</math> (<math>sp^3d^2</math> hybridization)</li> <li>• Pentagonal bipyramidal geometry (<math>\text{IF}_7</math>) (<math>sp^3d^3</math> hybridization)</li> <li>• Valence Shell Electron Pair Repulsion (VSEPR) Theory <math>\text{H}_2\text{O}</math>, <math>\text{ClF}_3</math>, <math>\text{ICl}_4^-</math></li> </ul>

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## Annual Teaching Plan

Academic Year: 2023-24 [Semesters: B.Sc. I (OE), Sem-I] Department: Chemistry


Subject: Chemistry Course Title: OEC03CHE11: General Aspects of Inorganic Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: July			Module/Unit:	Sub-units planned
Lectures	Practicals hr	Total	Induction	<ul style="list-style-type: none"> <li>• General Introduction</li> <li>• Discussion on Course Structure</li> <li>• Discussion on Examination pattern</li> <li>• Discussion on Syllabus</li> </ul>
06	-	06		
Month: August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Water	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Sources of water</li> <li>• Water quality parameters (definition, causes and estimation) like Colour, odour, pH, acidity, alkalinity.</li> </ul>
12	16	28		
Month: September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Water	<ul style="list-style-type: none"> <li>• Water quality parameters (definition, causes and estimation) like hardness, chlorides, total solids, dissolved oxygen, Chemical oxygen demand.</li> </ul>
12	16	28		
Month: October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Water	<ul style="list-style-type: none"> <li>• Ill effects of hard water in domestic and industrial purposes</li> <li>• Treatment of water by lime soda method, sterilization (chlorination),</li> <li>• Zeolite process, ion exchange process, reverse osmosis.</li> </ul>
12	16	28		

  
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## Annual Teaching Plan

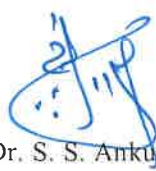
Academic Year: 2023-24 [Semesters: B.Sc. I (Major/Minor), Sem-II] Department: Chemistry

Subject: Chemistry

Course Title: DSC03CHE21: Physical Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals hr	Total	Chemical Energetics A) Thermodynamics	• Introduction, Spontaneous and nonspontaneous process with examples,
06	-	06		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemical Energetics A) Thermodynamics	• Statements of second law of thermodynamics
12	16	28		
Month: February			Module/Unit:	Sub-units planned
12	16	28	Chemical Energetics A) Thermochemistry	• Carnot's cycle and its efficiency
Month: March			Module/Unit:	Sub-units planned
09	16	25	Chemical Energetics A) Thermochemistry	• Problems based on Carnot Cycle



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Vivekanand College, Kolhapur (Empowered Autonomous)

Annual Teaching Plan

Academic Year: 2023-24

[Semesters: B.Sc. I (OE), Sem-II]

Department: Chemistry

Subject: Chemistry

Course Title: OEC03CHE21: General Aspects of Physical Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals hr	Total	Thermodynamics	• Introduction, Spontaneous and nonspontaneous process with examples,
06	-	06		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Thermodynamics	• Terms used in the study of thermodynamics
12	16	28		
Month: February			Module/Unit:	Sub-units planned
12	16	28	Thermodynamics	• Statements of second law of thermodynamics • Third Law of Thermodynamics

  
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## Annual Teaching Plan

Academic Year: 2023-24 [Semesters: B.Sc. I (IKS), Sem-II] Department: Chemistry

Subject: Chemistry

Course Title: IKS03GEC21: IKS

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals hr	Total	Chemistry of Ancient and Medieval India	• Introduction, Chemistry in pre-Harrapan period,
06	-	06		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemistry of Ancient and Medieval India	• Chemistry in Indus Valley civilization, Chemistry in post-Harrapan period,
12	16	28		
Month: February			Module/Unit:	Sub-units planned
12	16	28	Chemistry of Ancient and Medieval India	• Chemistry in Vedic and Ayurvedic: Chemistry in Charaka and Sushruta, Chemistry in Tantric Period: Rasaratnakara of Nagarjuna, Iatrochemical period.

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## Annual Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. II, Sem-III Department: Chemistry

Subject: Chemistry  
Chemistry

Course Title: DSC-1002D: Physical and Analytical & Industrial

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: July			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit Operations & Unit Processes	<ul style="list-style-type: none"> <li>• Introduction, Meaning of unit operations and processes &amp; its types.</li> </ul>
4	-	4		
Month: August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit Operations & Unit Processes	<ul style="list-style-type: none"> <li>• Distillation-Distillation of liquid mixtures, Types of distillation</li> <li>• Types of columns, packings and Condensers, Vacuum distillation, Spinning-band distillation and Steam distillation.</li> </ul>
4	-	04		
Month: September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit Operations & Unit Processes	<ul style="list-style-type: none"> <li>• Modes of Manufacturing: Batch, Semi-batch, Continuous</li> <li>• Study of Unit processes &amp; unit operations involved in manufacturing of ethanol by catalytic hydration of ethylene in vapour phase.</li> </ul>
4	-	04		
Month: October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total		
-	-	-		

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## Vivekanand College, Kolhapur (Empowered Autonomous)

### Annual Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. II, Sem-IV

Department: Chemistry

Subject: Chemistry

Course Title: DSC-1002D: Inorganic and Organic Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Crystal Field Theory	<ul style="list-style-type: none"><li>• Introduction</li><li>• Assumptions of CFT</li><li>• Crystal field stabilization energy (CFSE)</li><li>• Crystal field splitting of 'd' orbital in octahedral Complexes.</li></ul>
4	-	4		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Crystal Field Theory	<ul style="list-style-type: none"><li>• Comparison of CFSE for <i>Oh</i> and <i>Td</i> complexes</li><li>• Crystal field effects for weak and strong fields ligands, Tetrahedral symmetry,</li><li>• Crystal field splitting of 'd' orbital in Tetrahedral and square planar complex</li></ul>
4	-	04		
Month: February			Module/Unit:	Sub-units planned
4	-	04	Crystal Field Theory	<ul style="list-style-type: none"><li>• Jahn-Teller distortion,</li><li>• Limitations of CFT.</li><li>• Factors affecting the Magnitude of <math>10 Dq</math>, Spectrochemical series</li></ul>
Month: March			Module/Unit:	Sub-units planned
-	-	-	-	-

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## Annual Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. III, Sem-V

Department: Chemistry

Subject: Chemistry

Course Title: DSE-1002E1: Physical & Inorganic Chemistry  
DSE-1002E2: Organic & Analytical Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: July			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Bio-inorganic Chemistry	<ul style="list-style-type: none"> <li>• Introduction.</li> <li>• Essential and trace elements in biological process.</li> <li>• Metalloporphyrins with special reference to hemoglobin and myoglobin.</li> <li>• Biological role of alkali and alkaline earth metal ions with special reference to Na<sup>+</sup>, K<sup>+</sup> and Ca<sup>2+</sup></li> </ul>
04	-	04		
Month: August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Theory of Titrimetric Analysis	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Neutralization Indicators (Acid-Base Indicators)</li> <li>• Theory of indicators w.r.t. Ostwald's colour change interval and Ostwald's Quinoid theory</li> </ul>
08	28	60		
Month: September			Module/Unit:	Sub-units planned
08	28	60	Theory of Titrimetric Analysis	<ul style="list-style-type: none"> <li>• Neutralization curves and choice of indicators for the following titration, i. Strong acid-strong base ii. Strong acid-weak base iii. Strong base - weak acid</li> <li>• Complexometric titration: General account</li> <li>• Types of EDTA titration</li> </ul>
Month: October			Module/Unit:	Sub-units planned
08	28	60	Theory of Titrimetric Analysis	<ul style="list-style-type: none"> <li>• Metallochromic indicators w.r.t. Eriochrome Black-T indicator</li> </ul>
Month: November			Module/Unit:	Sub-units planned
03	14	16	Theory of Titrimetric Analysis	<ul style="list-style-type: none"> <li>• Redox titrations: General introduction, theory of redox indicators, Use of diphenyl amine and ferroin as redox indicators</li> </ul>

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## Annual Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. III, Sem-VI

Department: Chemistry

Subject: Chemistry

Course Title: DSE-1002F1: Physical & Inorganic Chemistry  
DSE-1002F2: Organic & Industrial Chemistry

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Corrosion and Passivity	<ul style="list-style-type: none"><li>• Introduction of corrosion</li><li>• Electrochemical theory of corrosion</li><li>• Factors affecting on corrosion, i. Position of metals in the electrochemical series on the basis of standard reduction potential ii. Purity of metal</li></ul>
06	-	06		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Corrosion and Passivity	<ul style="list-style-type: none"><li>• iii. Effect of moisture iv. Effect of oxygen</li><li>• Hydrogen overvoltage</li><li>• Methods of protections of metals from corrosion</li><li>• Passivity i. Definition ii. Types of passivity iii. Oxide film theory and evidences</li><li>• iv. Applications of passivity</li></ul>
8	28	60		
Month: February			Module/Unit:	Sub-units planned
8	28	60	Introduction to Chemical Industry	<ul style="list-style-type: none"><li>• General introduction, Indian Scenario of chemical industries, types of chemical industry, basic requirements of chemical industries, chemical production and raw materials; unit processes and unit operations and its types; modes of manufacturing-batch, semi-batch and continuous process;</li></ul>
Month: March			Module/Unit:	Sub-units planned
6	-	06	Introduction to Chemical Industry	<ul style="list-style-type: none"><li>• Introduction to various departments in industry: Quality control, Quality assurance, process development, Research and Development, Analytical development, Environmental health and safety.</li><li>• Industrial legislations-copy right act, patent act, trademarks; MSDS of hazardous chemicals.</li></ul>

Dr. S. S. Ankushrao



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## Annual Teaching Plan

Academic Year: 2023-24

Semesters: M.Sc. I, Sem-II

Department: Chemistry

Subject: Chemistry

Course Title: CC-1140 B: Analytical Chemistry-II

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Thermal Analysis	Introduction to thermal analysis, types of thermal analysis, significance of thermal analysis in Analytical Chemistry, effect of heat on materials, chemical decomposition, phase transformation etc. and general thermal analysis applications, advantages and disadvantages.
02	-	06		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Thermal Analysis	a) Thermogravimetry analysis (TGA): principle, instrumentation, working, types of TGA, factors influencing TGA, curve to show nature of decomposition reactions, the product and qualities of compounds expelled, TGA in controlled atmosphere, TGA curves, analysis, research and analytical implications of TGA.
04	-	12		
Month: February			Module/Unit:	Sub-units planned
04	-	12	Thermal Analysis	b) Differential thermal analysis (DTA) and differential scanning calorimetry (DSC), instrumentation, methodology, application and research implications.
Month: March			Module/Unit:	Sub-units planned
04	-	08	Thermal Analysis	Thermometric titrations method and applications
Month: April			Module/Unit:	Sub-units planned
02	-	06	Thermal Analysis	<b>Problems:</b> Simple problems based on TGA, DTA and DSC.

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## Annual Teaching Plan

Academic Year: 2023-24

Semesters: M.Sc. II, Sem-III

Department: Chemistry

Subject: Chemistry

Course Title: CC-1144 C: Advanced Spectroscopic Methods

Name of the Teacher: Dr. Sanjay Shivram Ankushrao

Month: July			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Proton NMR Spectroscopy	Part [A]: Recapitulation of proton NMR spectroscopy
02	-	02		
Month: August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Proton NMR Spectroscopy	Factors affecting coupling constants (Karplus curve variation, dihedral angle, bond order, electronegativity), analysis of First order spectra,
04	-	04		
Month: September			Module/Unit:	Sub-units planned
04	-	04	Proton NMR Spectroscopy	Complex spin-spin splitting of second order spectra, different spin systems (AB, AM, AX, ABX/AMX spin systems with examples). Simplification of complex spectra (High field strength, chiral resolving agent, effect of deuteration, nuclear magnetic double resonance.
Month: October			Module/Unit:	Sub-units planned
04	-	04	Proton NMR Spectroscopy	Shift reagent, solvent effect); Spectra of Homotopic, Enantiotopic and Diastereotopic systems. Part [B] Advanced NMR techniques (5) Fourier transform technique, nuclear overhauser effect (NOE),
Month: November			Module/Unit:	Sub-units planned
02	-	02	Proton NMR Spectroscopy	COSY, NOESY, Resonance of other nuclei – 19F, 31P.

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## Annual Teaching Plan

Academic Year: 2023-24 Semesters: B.Sc. I, Sem-II Department: Chemistry

Subject: Chemistry


Course Title: DSC-1002B: Physical and Analytical & Industrial Chemistry

Name of the Teacher: Mr. Satish Suresh Kadam

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemical Kinetics	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Terms in Chemical Kinetics</li> </ul>
4		4		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemical Kinetics	<ul style="list-style-type: none"> <li>• Factors affecting on Rate of reaction</li> <li>• Order and molecularity</li> <li>• First order reaction .</li> <li>• First order reaction Expression</li> </ul>
4	16	20		
Month: February			Module/Unit:	Sub-units planned
4	16	20	Chemical Kinetics	<ul style="list-style-type: none"> <li>• Second order reaction-Introduction</li> <li>• Derivaion of second order reaction</li> <li>• Examples</li> </ul>
Month: March			Module/Unit:	Sub-units planned
4	16	20	Chemical Kinetics	<ul style="list-style-type: none"> <li>• Characteristics of second order reactions</li> <li>• Numerical problems</li> </ul>
Month: April				
	8	8	Properties of liquids	<ul style="list-style-type: none"> <li>• Introduction, Classification of Physical properties,</li> <li>• Surface tension and chemical constitusion,</li> <li>• Viscosity</li> <li>• Refractive index</li> <li>• Numerical problems</li> </ul>

  
Mr. S. S. Kadam



  
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## Annual Teaching Plan

Academic Year: 2023-24 Semesters: B.Sc. II, Sem-IV Department: Chemistry

Subject: Chemistry Course Title: DSC-1002D: Inorganic and Organic

Chemistry Name of the Teacher: Mr. Satish Suresh Kadam

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Carboxylic acids and their derivatives	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Carboxylic acids (aliphatic and aromatic)</li> <li>• <i>Preparation</i>: Acidic and Alkaline hydrolysis of esters.</li> </ul>
4	-	4		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Carboxylic acids and their derivatives	<ul style="list-style-type: none"> <li>• <i>Reactions</i>: Hell -Vohlard - Zelinsky Reaction.</li> <li>• Carboxylic acid derivatives (aliphatic) :Preparation</li> <li>• preparation of Esters with mechanism</li> <li>• Comparative study of nucleophilicity of acyl derivatives</li> </ul>
4	16	20		
Month: February			Module/Unit:	Sub-units planned
4	16	20	Carboxylic acids and their derivatives	<ul style="list-style-type: none"> <li>• Reformatsky Reaction</li> <li>• Perkin condensation with mechanism and their applications.</li> </ul>
Month: March			Module/Unit:	Sub-units planned
4	16	20	Transition Elements	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• General group trends</li> <li>• Colour, magnetic and catalytic properties of transition elements.</li> </ul>
4	16	20	Transition Elements	<ul style="list-style-type: none"> <li>• Formation of complexes</li> <li>• Applications</li> </ul>



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# Vivekanand College, Kolhapur (Empowered Autonomous)

## Annual Teaching Plan

Academic Year: 2023-24 Semesters: B.Sc. III, Sem-VI Department: Chemistry

Subject: Chemistry Course Title: DSE-1002E1: Physical & Inorganic Chemistry  
DSE-1002E2: Organic & Industrial Chemistry

Name of the Teacher: Mr. Satish Suresh Kadam

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Name reactions	<ul style="list-style-type: none"> <li>• Introduction.</li> <li>• Beckmann,</li> <li>• Benzilic acid,</li> </ul>
04	-	04		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Name reactions	<ul style="list-style-type: none"> <li>• Baeyer Villiger,</li> <li>• Diels -Alder reaction,</li> <li>• Mannich Reaction,</li> <li>• Michael Reaction,</li> </ul>
4	28	32		
Month: February			Module/Unit:	Sub-units planned
4	28	32	Name reactions	<ul style="list-style-type: none"> <li>• Fries,</li> <li>• Dienone-Phenol rearrangement</li> <li>• Problems based on reactions.</li> </ul>
Month: March			Module/Unit:	Sub-units planned
4	28	32	Electrophilic Addition to $>C=C<$ and $-C\equiv C-$ bonds	<ul style="list-style-type: none"> <li>• Addition to Carbon-Carbon double (<math>&gt;C=CC=C&lt;</math> bond)</li> <li>• Ozonolysis (formation of aldehydes &amp; ketones).</li> <li>• Addition to Carbon-Carbon triple (<math>-C\equiv C-</math>) bond:</li> </ul>
Month: April			Module/Unit:	Sub-units planned
03	-	03	Electrophilic Addition to $>C=C<$ and $-C\equiv C-$ bonds	<ul style="list-style-type: none"> <li>• Mechanism of electrophilic addition to <math>-C\equiv C-</math> bond.</li> </ul>

Mr. S. S. Kadam

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Annual Teaching Plan

Academic Year: 2023-24 Semesters: B. Sc. I (Maj+Min), Sem-II Department: Chemistry

Subject: Chemistry Course Title: Chemistry Practicals

Name of the Teacher: Dr. Mrs. Sarita Dattajirao Shinde

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>Determination of enthalpy of hydration of copper sulphate.</li> <li>Study of the solubility of benzoic acid in water and determination of <math>\Delta H</math>.</li> <li>Determination of enthalpy of neutralization of HCl with NaOH</li> </ul>
-	16	16		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>To study the reaction rate of hydrolysis of methyl acetate in presence of 0.5N HCl.</li> <li>To determine viscosity of given liquid A and B.</li> <li>To determine equivalent weight of Mg by Eudiometer.</li> <li>Estimation of Aniline</li> </ul>
-	16	16		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li><b>Spot Tests</b> Detection of following cations using spot tests: <math>\text{Cu}^{2+}</math>, <math>\text{Co}^{2+}</math>, <math>\text{Ni}^{2+}</math>, <math>\text{Fe}^{3+}</math>, <math>\text{Al}^{3+}</math>, <math>\text{Zn}^{2+}</math>, <math>\text{Mg}^{+2}</math>, <math>\text{Pb}^{2-}</math></li> <li><b>Paper Chromatography</b> Detection of following cations using Paper Chromatography: <math>\text{Cu}^{2+}</math>, <math>\text{Co}^{2+}</math>, <math>\text{Co}^{2+}</math>, <math>\text{Ni}^{2+}</math>, <math>\text{Ni}^{2+}</math>, <math>\text{Cu}^{2+}</math></li> <li>To estimate amount of Cu (II) ions by iodometric titration by using <math>\text{Na}_2\text{S}_2\text{O}_3</math> solution</li> </ul>
-	16	16		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>Organic Spotting</li> </ul>
-	4	4		
Month: April-May				Practical and Theory Examination

  
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Annual Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. II, Sem-IV

Department: Chemistry

Subject: Chemistry

Course Title: DSC-1002D: Inorganic Chemistry

Name of the Teacher: Dr. Mrs. Sarita Dattajirao Shinde

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Coordination Chemistry: Theories of Metal Complexes A] Valence Bond Theory	<ul style="list-style-type: none"> <li>• Definition and formation of co-ordinate covalent bond in <math>\text{BF}_3\text{-NH}_3</math> and <math>[\text{NH}_4]^+</math>,</li> <li>• Distinguish between double salt and complex salt, Werner's theory i) Postulates, ii) theory as applied to cobalt amines complexes;</li> <li>• Description of the terms: ligands, co-ordination compounds. Coordination number;</li> <li>• IUPAC system of nomenclature,</li> <li>• Preparation of methyl orange</li> <li>• Preparation of p-nitro acetanilide</li> </ul>
8	16	24		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Coordination Chemistry: Theories of Metal Complexes A] Valence Bond Theory	<ul style="list-style-type: none"> <li>• Postulates of VBT, Inner and outer orbital complexes w. r. t. coordination numbers 4 and 6; Drawbacks of VBT.</li> <li>• Structural and stereoisomerism in complexes with coordination numbers 4 and 6;</li> <li>• Estimate the amount of metal present in a given solution gravimetrically-Ni as Ni-DMG, Ba as <math>\text{BaSO}_4</math></li> <li>• Determination of the surface tension of a liquid or a dilute solution using a Stalagmometer.</li> <li>• Study of the variation of surface tension of a detergent solution with concentration</li> </ul>
8	16	24		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Coordination Chemistry: Theories of Metal Complexes B] Crystal Field Theory	<ul style="list-style-type: none"> <li>• Assumptions of CFT, Crystal field splitting of 'd' orbital in octahedral,</li> <li>• tetrahedral and square planar complex, Crystal field stabilization energy (CFSE),</li> <li>• Determination of the relative and absolute viscosity of a liquid or dilute solution using an Ostwald's viscometer.</li> <li>• To investigate the reaction between potassium per sulphate and KI (Equal Concentration)</li> </ul>
8	16	24		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Coordination Chemistry: Theories of Metal Complexes B] Crystal Field Theory	<ul style="list-style-type: none"> <li>• Comparison of CFSE for Oh and Td complexes, Crystal field effects</li> <li>• for weak and strong fields ligands, Tetrahedral symmetry, Factors affecting</li> <li>• the Magnitude of <math>10 Dq</math>, Spectrochemical series, Jahn-Teller distortion,</li> <li>• Limitations of CFT</li> <li>• Estimation of Acetone</li> <li>• Estimations of Vitamin-C from tablets</li> </ul>
8	16	24		

Month: April - May			Module/Unit:	Sub-units planned
-	-	-		<ul style="list-style-type: none"> <li>Practical and Theory Examination</li> </ul>

*S. D. Shinde*

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Annual Teaching Plan

Academic Year: 2023-24

Semesters: B. Sc. III, Sem-VI

Department: Chemistry


Subject: Chemistry

Course Title: 1002F2: Inorganic Chemistry


Name of the Teacher: Dr. Mrs. Sarita Dattajirao Shinde

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Coordination Chemistry A) Inorganic Reaction mechanism	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Classification of Mechanism</li> <li>• Study of Mechanism Association, dissociation, interchange and the rate determining steps</li> <li>• SN<sup>1</sup> and SN<sup>2</sup> reaction for inert and labile complexes</li> <li>• Mechanism of substitution in cobalt (III) octahedral complexes</li> <li>• Trans effect and its theories</li> <li>• Applications of trans effect in synthesis of Pt (II) complexes</li> <li>• Determination of percentage of magnesium in the given sample of talcum powder.</li> <li>• Determination of amount of aluminum in the given solution of potash alum.</li> </ul>
8	28	36		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Coordination Chemistry B) Thermodynamic and Kinetic aspects of metal complexes	<ul style="list-style-type: none"> <li>• Introduction, Thermodynamic stability and Kinetic Stability,</li> <li>• Relation between thermodynamic and kinetic stability, Stepwise stability constant,</li> <li>• Factor affecting the stability of complexes,</li> <li>• Determination of Stability constant by Job variation, Mole ratio and Slope ratio method.</li> <li>• Determination of titrable acidity in the given sample of milk or lassi.</li> <li>• Determination of Fe in cement.</li> </ul>
8	28	36		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemistry of f-Block Elements A) Lanthanides	<ul style="list-style-type: none"> <li>• Introduction, Occurrence,</li> <li>• Electronic Configuration, Oxidation State, Lanthanide contraction,</li> <li>• Separation of Lanthanides by Ion exchange method.</li> <li>• Determination of chlorine in bleaching powder.</li> <li>• Determination of Free acidity in ammonium sulphate fertilizer.</li> <li>• Determination of percentage of CaCO<sub>3</sub> in chalk.</li> <li>• Determination of COD.</li> </ul>
8	14	22		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemistry of f-Block Elements B) Actinides	<ul style="list-style-type: none"> <li>• Position in periodic table, Electronic configuration,</li> <li>• General methods of preparation of transuranic elements - i. Neutron capture - followed by β decay, ii. Accelerated projectile bombardment, iii. Heavy ion bombardment,</li> </ul>
6	-	06		

				<ul style="list-style-type: none"><li>• IUPAC nomenclature of the super heavy elements with atomic number (Z) greater than 100.</li></ul>
Month : April - May				<ul style="list-style-type: none"><li>• Practical and theory Examination</li></ul>

  
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Annual Teaching Plan

Academic Year: 2023-24

Semesters: M. Sc. I, Sem-II

Department: Chemistry

Subject: Inorganic Chemistry

Course Title: DSE14CHE21: Applied Inorganic Chemistry

Name of the Teacher: Dr. Mrs. Sarita Dattajirao Shinde

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Metal Compounds in Medicine	<ul style="list-style-type: none"> <li>Medicinal use of metal complexes as antibacterial, antiviral, antibiotics &amp; related compounds.</li> </ul>
3	-	3		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Metal Compounds in Medicine	<ul style="list-style-type: none"> <li>Metal deficiency and diseases: iron, zinc and copper deficiency,</li> <li>Metals used for diagnosis</li> </ul>
4	-	4		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Metal Compounds in Medicine	<ul style="list-style-type: none"> <li>chemotherapy with particular reference to anti cancer drugs,</li> <li>Chelate therapy, chemotherapy with compounds of some non essential elements;</li> </ul>
4	-	4		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Metal Compounds in Medicine	<ul style="list-style-type: none"> <li>Lithium in psycho pharmacological drugs,</li> <li>Antimicrobial agents.</li> <li>use of cis-platin as antitumor drug,</li> <li>Gold containing drugs used in the therapy of Rheumatic-Arthritis and cancer,</li> </ul>
4	-	4		

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## Vivekanand College, Kolhapur (Autonomous)

### Annual Teaching Plan

Academic Year: 2023-24

Semesters: B.Sc. III, Sem-V

Department: Chemistry

Subject: Chemistry

Course Title: DSE-1002E1: Physical Chemistry

Name of the Teacher: Dr. Asmita Shashikant Tapase

Month: June			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Quantum Theory	<ul style="list-style-type: none"> <li>• Introduction,</li> <li>• Dual nature of matter and energy: De Broglie hypothesis</li> </ul>
02	-	02		
Month: July			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Quantum Theory	<ul style="list-style-type: none"> <li>• The Heisenberg's uncertainty principle,</li> <li>• Concept of energy operators (Hamiltonian)</li> <li>• Derivation of Schrodinger wave equation, Physical interpretation of <math>\psi</math> and <math>\psi^2</math>,</li> <li>• Particle in a one dimensional box, Schrodinger wave equation for hydrogen atom, Concept of quantum numbers.</li> </ul>
08	21	29		
Month: August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Quantum Theory	<ul style="list-style-type: none"> <li>• Particle in a one dimensional box,</li> <li>• Schrodinger wave equation for hydrogen atom</li> <li>• Concept of quantum numbers.</li> </ul>
08	28	36		
			Photochemistry	<ul style="list-style-type: none"> <li>• Introduction –</li> <li>• Difference between thermal and photochemical processes.</li> <li>• Laws of photochemistry:</li> <li>• i) Grothus - Draper law,</li> <li>• ii) Lambert law,</li> <li>• iii) Lambert-Beer's law (with derivations),</li> <li>• iv) Stark-Einstein law</li> </ul>
Month: September			Module/Unit:	Sub-units planned
08	28	36	Photochemistry	<ul style="list-style-type: none"> <li>• Quantum yield,</li> <li>• reasons for high and low quantum yield,</li> <li>• Factors affecting Quantum yield,</li> <li>• Photosensitized reactions–dissociation of <math>H_2</math>, photosynthesis,</li> <li>• Examples of Photochemical recations-                             <ol style="list-style-type: none"> <li>1)Photo-dimerization of anthracene,</li> <li>2) decomposition of HI and HBr,</li> </ol> </li> <li>• Photophysical and photochemical processes</li> </ul>

Month: October			Module/Unit:	Sub-units planned
5	28	5	Photochemistry	<ul style="list-style-type: none"> <li>• Jablonaski diagram depicting various processes occurring in the excited state:</li> <li>• Qualitative description of fluorescence and phosphorescence,</li> <li>• Chemiluminescence,</li> <li>• Electroluminescence,</li> <li>• Numerical problems</li> </ul>
Month : November			-	<ul style="list-style-type: none"> <li>• Theory Examination</li> </ul>



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**Vivekanand College, Kolhapur (Empowered Autonomous)**

**Annual Teaching Plan**

**Academic Year: 2023-24**

**Semesters: B. Sc. II, Sem-II Department: Chemistry**

**Subject: Chemistry**

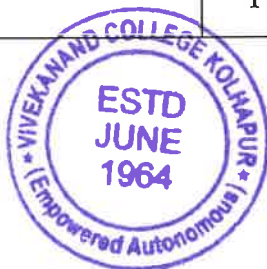
**Course Title: Chemistry Practicals**

**Name of the Teacher: Dr. Asmita Shashikant Tapase**

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>To determine the normality strong acid and weak acid by titrating against strong base conductometrically</li> <li>To determine the normality strong acid by titrating against strong base potentiometrically</li> <li>To investigate the reaction between potassium per sulphate and KI (equal conc.)</li> <li>To investigate the reaction between potassium per sulphate and KI (unequal conc.)</li> </ul>
-	16	16		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>Preparation of Ferrous ammonium sulphate</li> <li>Preparation of potash alum</li> <li>Preparation of Tetraamine copper sulphate</li> <li>Preparation of p-nitro aniline</li> </ul>
-	16	16		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>Semi micro qualitative analysis (2 cations, 2 anions)</li> </ul>
-	16	16		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	<ul style="list-style-type: none"> <li>Organic Spotting</li> </ul>
-	4	4		
Month: April-May				Practical and Theory Examination

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**Annual Teaching Plan**

**Academic Year: 2023-24    Semesters: B. Sc. I, Sem-II (OE)    Department: Chemistry**

**Subject: Chemistry                      Course Title: General Aspects of Physical Chemistry**

**General Aspects of Analytical Chemistry**

**Name of the Teacher: Dr. Asmita Shashikant Tapase**

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Nuclear Chemistry	<ul style="list-style-type: none"> <li>• Determination of enthalpy of hydration of copper sulphate.</li> <li>• Study of the solubility of benzoic acid in water and determination of <math>\Delta H</math>.</li> <li>• Determination of enthalpy of neutralization of HCl with NaOH</li> </ul>
04	-	04		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Nuclear Chemistry	<ul style="list-style-type: none"> <li>• To study the reaction rate of hydrolysis of methyl acetate in presence of 0.5N HCl.</li> <li>• To determine viscosity of given liquid A and B.</li> <li>• To determine equivalent weight of Mg by Eudiometer.</li> <li>• Estimation of Aniline</li> </ul>
04	16	16		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Environmental Chemistry	<ul style="list-style-type: none"> <li>• Concept and scope of Environmental Chemistry</li> <li>• Environmental segments</li> <li>• The nature of cycles of the environment-general introduction of Hydrogen Cycle, Oxygen Cycle, Nitrogen Cycle,</li> <li>• Phosphate Cycle, Sulphur cycle</li> </ul>
04	16	16		
Month: March			Module/Unit:	Sub-units planned

Lectures	Practicals	Total	Environmental Chemistry	<ul style="list-style-type: none"> <li>• Detail explanation of Oxygen Cycle</li> <li>• Composition of the atmosphere</li> <li>• Toxic trace elements in the natural water and waste water</li> <li>• Air pollutants</li> <li>• Chemistry of acid rain</li> </ul>
04	4	4		
Month: April-May				Practical and Theory Examination

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**Annual Teaching Plan**

**Academic Year: 2023-24      Semesters: B. Sc. III, Sem-VI      Department: Chemistry**

**Subject: Chemistry      Course Title: Physical Chemistry**

**Name of the Teacher: Dr. Asmita Shashikant Tapase**

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	▪ Adsorption	<ul style="list-style-type: none"> <li>• Introduction, Adsorption as a surface phenomenon (mechanism),</li> <li>• Definition of important basic terms: absorption, adsorption, adsorbant, adsorbate, interface etc.,</li> <li>• Distinction between adsorption and absorption,</li> <li>• Characteristics of adsorption,</li> <li>• Factors affecting adsorption, Types of adsorption,</li> <li>• Distinction between physical adsorption and chemical adsorption, Adsorption isotherms: Freundlich, Langmuir adsorption isotherm, BET equation (derivation not expected), determination of surface area using Langmuir method and BET equations</li> </ul>
08	28	36		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	▪ Thermodynamics	<ul style="list-style-type: none"> <li>• Introduction.</li> <li>• Free energy: Gibbs function (G) and Helmholtz function (A), Criteria for thermodynamic equilibrium and spontaneity.</li> <li>• Relation between <math>\Delta G</math> and <math>\Delta H</math> : Gibbs-Helmholtz equation.</li> <li>• Phase equilibria : Clapeyron – Clausius equation and its applications.</li> </ul>
08	28	36		

				Thermodynamic derivation of law of mass action, Van't – Hoff isotherm and isochore
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	■ Thermodyna mics  Chemical Kinetics -	<ul style="list-style-type: none"> <li>• Thermodynamic derivation of law of mass action, Van't – Hoff isotherm and isochore.</li> <li>• Fugacity and activity concepts.</li> <li>• Partial molar quantities, Partial molar volume, Concept of chemical potential, GibbsDuhem equation.</li> <li>• Numerical problems</li> <li>• Introduction</li> <li>• Third Order Reaction -Derivation of rate constant and properties and examples.</li> <li>• Simultaneous reactions such as</li> <li>• Opposing reaction: (Derivation of rate equation for first order opposed by first order expected)</li> </ul>
08	28	36		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Chemical Kinetics -	i. Side reaction.  ii. Consecutive reactions.  iii.Chain reaction.  iv.Explosive reaction (Derivation of rate equation )
04	14	18		
Month: April-May				Practical and Theory Examination

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**Vivekanand College, Kolhapur (Empowered Autonomous)**

**Annual Teaching Plan**

**Academic Year: 2023-24      Semesters: M. Sc. II, Sem-IV      Department: Chemistry**

**Subject: Chemistry      Course Title: Inorganic Chemistry**

**Name of the Teacher: Dr. Asmita Shashikant Tapase**

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	▪ Applications of nanomaterials -	<ul style="list-style-type: none"> <li>• Carbon nanomaterials</li> <li>• Nanocomposites include metal nanomaterials such as single particle as well as</li> </ul>
04	-	36		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	▪ Applications of nanomaterials -	<ul style="list-style-type: none"> <li>• coreshell nanomaterials. Fuel cell, Solar cell, medicinal applications, agro-food applications</li> <li>• c) Polymer Nanotechnology</li> </ul>
04	-	04		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	▪ Applications of nanomaterials	<ul style="list-style-type: none"> <li>• Organic Electronics</li> <li>• Nanotribology</li> </ul>
04	-	04		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	▪ Applications of nanomaterials -	<ul style="list-style-type: none"> <li>• Nanobiotechnology</li> </ul>
03	-	03		
Month: April-May				Practical and Theory Examination

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# Vivekanand College, Kolhapur (Autonomous)

## Syllabus Completion Report

Academic Year - 2023-24

Sem-I, Sem-III, Sem-V

Department - Chemistry

Name of the Teacher – **Dr. A. S. Tapase**

Name of the Class	Units Allotted	Units Completed	Remark
B.Sc. I	-	-	-
B.Sc. II	<ul style="list-style-type: none"><li>Chemical Kinetic</li><li>Electrochemistry I</li><li>Conductometric Titrations</li></ul>	<ul style="list-style-type: none"><li>Chemical Kinetics</li><li>Electrochemistry I</li><li>Conductometric Titrations</li></ul>	Completed
B.Sc. III	<ul style="list-style-type: none"><li>Quantum Theory</li><li>Photochemisry</li></ul>	<ul style="list-style-type: none"><li>Quantum Theory</li><li>Photochemisry</li></ul>	Completed
M.Sc. II	<ul style="list-style-type: none"><li>Magnetic Materials</li></ul>	<ul style="list-style-type: none"><li>Magnetic Materials</li></ul>	Completed
M.Sc. II	<ul style="list-style-type: none"><li>Surface characterization by spectroscopy</li></ul>	<ul style="list-style-type: none"><li>Surface characterization by spectroscopy</li></ul>	Completed



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## Syllabus Completion Report

Academic Year - 2023-24

Sem-II, Sem-VI, Sem-IV


Department - Chemistry

Name of the Teacher – **Dr. A. S. Tapase**

Name of the Class	Units Allotted	Units Completed	Remark
B.Sc. I (OE) General aspects of Physical Chemistry	✚ Nuclear Chemistry	✚ Nuclear Chemistry	Completed
B.Sc. I (OE) General aspects of Applied Chemistry	✚ Environmental Chemistry	✚ Environmental Chemistry	Completed
B.Sc. III	▪ Adsorption ▪ Thermodynamics ▪ Chemical Kinetics	▪ Adsorption ▪ Thermodynamics ▪ Chemical Kinetics	Completed
M.Sc. II (Inorganic Chemistry)	▪ Applications of nanomaterials	▪ Applications of nanomaterials	Completed



Dr. A. S. Tapase

  
Dr. Mrs. S. D. Shirke  
**HEAD**  
DEPARTMENT OF CHEMISTRY  
VIVEKANAND COLLEGE, KOLHAPUR  
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Vivekanand College, Kolhapur (Empowered Autonomous)

Annual Teaching Plan

Academic Year - 2023-24 (Sem II & IV)

Department- Chemistry

Name of the Teacher – Dr. D. S. Gaikwad

Month – December				
M.Sc. I Sem. II (NEP):- Course Title:- Organic Chemistry-II				
Lectures	Practicals	Total	Module Unit	Sub-Units Planned
06	--	06	Study of following reactions	a) Study of following reactions Mechanism of condensation reaction involving enolates, Arndt-Eistert, Stobbe, Simon-smith, Ulmann reaction.
M.Sc.II Sem IV :- Course Title:- Organic Chemistry				
Lectures	Practicals	Total	Module Unit	Sub-Units Planned
06	32	38	Vitamins	Introduction of Vitamins, Classification and nomenclature of Vitamins, Sources of vitamins and their deficiency, Synthesis, structure.

Month – January				
M.Sc.I Sem. II:- Course Title:- Organic Chemistry-II				
Lectures	Practicals	Total	Module Unit	Sub-Units Planned
08	--	08	Study of following reactions	Study of following reactions and their applications: McMurry, Prins, Nef, Passerini, Baylis-Hilman, Mitsunobu reaction.
M.Sc.II Sem IV :- Course Title:- Organic Chemistry				
Lectures	Practicals	Total	Module Unit	Sub-Units Planned
12	48	60	Vitamins	Biological functions of vitamin B1, B2, B5, B6 and Biotin (Vitamin H).
			Alkaloids	Introduction, occurrence, isolation and functions of alkaloids, Structure, stereochemistry and synthesis of the following: Morphine, Reserpine.
M.Sc.II Sem IV :- Course Title:- Analytical Chemistry				
04	--	04	Analysis of Paint, Pigment and Petroleum products	Composition of paint, Preliminary inspection of sample, test on the total coating, separation and estimation of pigments, Binder and thinner of latex paints.
B.Sc. III Sem VI Inorganic Chemistry				

03	--	03	Iron and Steel	Occurrence and ores of iron, Definition of the terms-Ore, Minerals, Slag, Flux, Gangue, Matrix, Calcination.
<b>Month – February</b>				
<b>M.Sc.I Sem. II (NEP):- Course Title:- Organic Chemistry-II</b>				
: Lectures	Practicals	Total	Module Unit	Sub-Units Planned
08	--	08	Study of following reactions	Study of following reactions and their applications: Darzen, Duff, Click reaction.
<b>M.Sc.II Sem IV :- Course Title:- Organic Chemistry</b>				
Lectures	Practicals	Total	Module Unit	Sub-Units Planned
16	32	48	Stereochemistry	Stereochemistry of compounds containing no chiral carbon atoms and diastereoisomerism (Geometrical isomerism). a) Stereochemistry of Allenes, Spiranes and Biphenyls
			Alkaloids	Introduction, occurrence, isolation and functions of alkaloids, Structure, stereochemistry and synthesis of the following: Atropine and Conin.
<b>M.Sc.II Sem IV :- Course Title:- Analytical Chemistry</b>				
04	--	04	Analysis of Paint, Pigment and Petroleum products	Flash points, Practical applications and examples in analytical chemistry and research. Introduction to petroleum products, quality control, Specific gravity, Viscosity, cloud point, pour point, flash point, vapour pressure. Calorific value by bomb calorimeter.
<b>B.Sc. III Sem VI Inorganic Chemistry</b>				
03	--	03	Iron and Steel	Reduction, Roasting, Smelting and Leaching, Extraction of iron by Blast furnace. Steel: Definition and types, Heat treatment on steel

<b>Month – March</b>				
<b>M.Sc.II Sem IV :- Course Title:- Analytical Chemistry</b>				
: Lectures	Practicals	Total	Module Unit	Sub-Units Planned
04	--	04	Analysis of Paint, Pigment and Petroleum products	Doctor test, sulphuric acid absorption, aniline point and colour determination. Determination of water, neutralization value, ash content, sulphur and mercaptan sulphur. Determination of lead in petroleum.

M.Sc.II Sem IV :- Course Title:- Organic Chemistry				
Lectures	Practicals	Total	Module Unit	Sub-Units Planned
12	24	36	Stereochemistry	Assignment of configuration b) Configuration of diastereomers (Geometrical isomerism) based on physical and chemical methods.
B.Sc. III Sem VI Inorganic Chemistry				
02	--	02	Iron and Steel	Conversion of Cast iron into steel by Bessemer process and LD process. Heat treatment on steel.

*Dipar*  
Dr. D. S. Gaikwad



*S.D. Shirke*  
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Vivekanand College, Kolhapur (Empowered Autonomous)

Annual Teaching Plan

Academic Year - 2023-24

Department- Chemistry (Sem I & III)

Name of the Teacher – Dr. A. A. Patravale

Month – July				
M.Sc.II Sem III :- Course Title:- Organic Chemistry (Paper: Drugs and heterocycles)				
Lectures	Practicals	Total	Unit	Subunit planned
08	24	32	Drug Design	Development of new drugs, procedures followed in drug design. History and development of Quantitative structure activity relationship (QSAR). Concepts of drug receptors, Relation of chemical structure and chemical activity.
Month – August				
M.Sc.I Sem I (NEP):- Course Title:- Organic Chemistry (Optional elective)				
Lectures	Practicals	Total	Unit	Sub-Units Planned
08	--	08	Aromatic Electrophilic Substitutions	Introduction, the arenium ion mechanism, orientation and reactivity in Nitration, Sulphonation, Friedel-Crafts and Halogenation in aromatic systems, energy profile diagrams. The ortho/para ratio, ipso attack, orientation in their ring systems. Diazo-coupling, Vilsmeier Haak reaction, Nucleophilic aromatic substitution reactions SN1, SN2.
M.Sc.II Sem III :- Course Title:- Organic Chemistry (Paper: Drugs and heterocycles)				
Lectures	Practicals	Total	Unit	Sub-Units Planned
04	---	58	Study of Antibiotics	Introduction, $\beta$ -lactum Antibiotics, cephalosporin Antibiotics, SAR of $\beta$ -lactum and cephalosporin, Structural features of tetracycline & macrocyclic antibiotics (no synthesis).
--	54	---	Practical	Major Organic Experiment Minor Organic Experiment
M.Sc. II Sem III :- Course Title:- Organic Chemistry (Paper: Drugs and heterocycles)				
Lectures	Practicals	Total	Unit	Sub-Units Planned
08	--	08	Study of	Antimalerials: Trimethoprim.

				Following types of Drug	b) Analgesic & Antipyretics: Paracetamol, Meperidine, methadone, Aminopyrine. c) Anti-inflammatory: Oxyphenylbutazone, Diclophenac, Indomethacin.
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**Month – September**

**M.Sc.I Sem II :- Course Title:- Organo Analytical Chemistry**

Lectures	Practicals	Total	Unit	Sub-Units Planned
03	--	03	Clinical Analysis	Biological significance, analysis of assay of enzyme (pesine, monoamine, oxidase, tyrosinase,). Composition and detection of some abnormal level of certain constituent leading to diagnosis of diseases, Sample collection and preservation of body fluid.

**M.Sc.II Sem III :- Course Title:- Organic Chemistry (paper: Drugs and Heterocycles)**

Lectures	Practicals	Total	Unit	Subunit planned
12	--	44	Study of Following types of Drug	d) Antitubercular & antileprotic: Dapsone e) Anaesthetics: Lidocaine, Thiopental. f) Antihistamines: Diphenylhydramine. g) Tranquilizers: Diazepam, Trimeprazine.

**M.Sc.I Sem I (NEP):- Course Title:- Organic Chemistry Practicals**

Lectures	Practicals	Total	Unit	Sub-Units Planned
--	32	--	Practical	Major Organic Experiment Minor Organic Experiment

**M.Sc.I Sem I (NEP):- Course Title:- Organic Chemistry (Optional elective)**

Lectures	Practicals	Total	Unit	Sub-Units Planned
03	--	03	Non benzenoid aromatic Compounds	Aromaticity in Non- benzenoids compounds Annulenes and heteroannulenes,

**Month – October**

**M.Sc.I Sem II :- Course Title:- Organo Analytical Chemistry**


Lectures	Practicals	Total	Unit	Sub-Units Planned
03	--	03	Clinical Analysis	Analytical method of detection of body fluid, Estimation of Glucose, Cholesterol, Urea, uric acid, Creatinine, Calcium, Phosphate, Sodium, Potassium, Chloride

**M.Sc.II Sem III :- Course Title:- Organic Chemistry**

Lectures	Practicals	Total	Unit	Sub-Units Planned
12	--	36	Study of	h) Anti AIDS: General study Introduction, structure and life cycle of the

			Following types of Drug	AIDS virus, recent development, Azedothymidine (AZT) derivatives j) <b>Anti-neoplastic drugs:</b> Introduction, Cancer chemotherapy, Synthesis of mechloraethamine, cyclophosphamide, Mephalan, uracils, mustards. Recent development in cancer chemotherapy. Hormones and natural products.
<b>M.Sc.I Sem I (NEP):- Course Title:- Organic Chemistry Practicals</b>				
--	24	--	Practical	Major Organic Experiment Minor Organic Experiment
<b>M.Sc. II Sem III :- Course Title:- Analytical Chemistry (Organoanalytical chemistry)</b>				
Lectures	Practicals	Total	Unit	Sub-Units Planned
03	--	03	Study of following types of Drugs	i) <b>Cardiovascular:</b> Synthesis of dilliazem, quinidine, methyldopa, atenolol, oxyprenol.
Month – November				
<b>M.Sc.II Sem III :- Course Title:- Organic Chemistry</b>				
Lectures	Practicals	Total	Unit	Sub-Units Planned
02	--	02	Organic practical	Minor Organic Experiment
<b>M.Sc.I Sem I (NEP):- Course Title:- Organic Chemistry (Optional elective)</b>				
Lectures	Practicals	Total	Unit	Sub-Units Planned
02	--	02	Non Benzenoides Compounds	fulvene, tropylium salts, ferrocene

  
Dr. A. A. Patravale

  
Dr. Mrs. S. D. Shirke  
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**Vivekanand College, Kolhapur (Autonomous)**  
**Annual Teaching Plan**  
 Academic Year – 2023-24, M.Sc Sem. II, IV  
 Department- Chemistry  
 Name of the Teacher – **Dr. A. A. Patravale**


Month – December				
M.Sc.I Sem. II:- Course Title:- Analytical Chemistry-II				
: Lectures	Practical's	Total	Module Unit	Sub-Units Planned
06	--	06	Practical	Estimation of various drug and chemicals
M.Sc.II Sem IV :- Course Title:- Organic Chemistry				
Lectures	Practical's	Total	Module Unit	Sub-Units Planned
12	32	44	Benzenoid and Non benzenoid aromatic Compounds	a) Polycyclic aromatic compounds: Synthesis, reactions, Linear and non-linear ortho fused polynuclear hydrocarbons.

Month – January				
M.Sc.I Sem. II:- Course Title:- Organic Chemistry-II				
: Lectures	Practical's	Total	Module Unit	Sub-Units Planned
08	--	08	a) Oxidation	Oxidation of alcohol to aldehyde, ketone or acid: Jones reagent, Swern oxidation, Collins reagent, Fetizones reagent, PCC, PDC, IBX, Activated MnO <sub>2</sub> , Chromyl chloride (Etard reaction), TEMPO, NMO, Moffatt oxidation.
M.Sc.II Sem IV :- Course Title:- Industrial Analytical Chemistry				
Lectures	Practical's	Total	Module Unit	Sub-Units Planned
12	48	60	Analysis of Soil and Fertilizer	Introduction, Method of Soil analysis, Soil fertility, determination of inorganic constitution of plant,

Month - February				
M.Sc.I Sem. II:- Course Title:- Organic Chemistry-II				
Lectures	Practical's	Total	Module Unit	Sub-Units Planned
08	--	08	b) Hydroboration c) Enamines	Mechanism and Synthetic Applications Formation and reactivity of enamines
M.Sc.II Sem IV :- Course Title:- Analytical Chemistry				
Lectures	Practical's	Total	Module Unit	Sub-Units Planned
16	32	48	Synthesis and applications of perfumery	Introduction to perfumery compounds and its commercial process, essential oil, method of preparation and importants, synthesis of 2-Phenylethanol, Yara-yara, vanillin
				preparation and importants, synthesis of other food flavours, synthetic musk, Jasmone, ionones, beta-ionones from citral, phenyl acetic acid and its ester, benzy acetate.

Month - March				
M.Sc.I Sem. II:- Course Title:- Organic Chemistry-II				
Lectures	Practical's	Total	Module Unit	Sub-Units Planned
08	--	08	Analysis of Soil and Fertilizer	Classification of fertilizer, clinical Analysis as measure of soil fertility, fertilizer and its detection method.
M.Sc.II Sem IV :- Course Title:- Analytical Chemistry				
Lectures	Practical's	Total	Module Unit	Sub-Units Planned
12	24	36	Dyes and Intermediates	Classification and synthesis of important dye intermediates by using nitration, sulphonation, diazotization reactions. Commercial processes for azo-dyes, reactive dyes, optical brighteners, thermal sensitive dyes, dispersed dyes and reactive dyes.

  
Dr. A. A. Patravale

  
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# Vivekanand College, Kolhapur (Empowered Autonomous)

Department of Chemistry

Academic Year: 2023-24

## Annual Teaching Plan

Name of the teacher: Dr. S. D. Kharade

Programme: M.Sc. I Semester II

Subject: Chemistry

Course Title: Inorganic Chemistry

Month December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	Theory and Practical Examination of previous term
-	-	-		
Month January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-II A) Stereochemistry and bonding in main group compounds	Introduction to VSEPR theory and drawbacks, bond length, bond angles, bond energies and resonance, Structure and bonding in some inorganic molecules,
4	-	4		
Month February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-II A) Stereochemistry and bonding in main group compounds	some simple reactions of covalently bonded molecules (atomic inversion, Berry pseudorotation, nucleophilic displacement, and free radical reaction).
4	-	4		
Month March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-II A) Stereochemistry and bonding in main group compounds	P $\pi$ - P $\pi$ and P $\pi$ - d $\pi$ bonds, Bent rule, walsh diagram, back bonding. Stability constant, step wise and overall formation constant and their interaction, trends in stepwise constants, factors affecting the stability of
2	-	2		
2	-	2	B) Metal - ligand equilibria in solution	

				metal complexes with reference to the nature of metal ion and ligand, chelate effect, ternary complexes and factors affecting their stabilities,
<b>Month April</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-II	Stability of metal complexes of crown ether, determination of stability constant for binary complexes using pH-metric (Bjerrums method) and spectrophotometric (Job's and mole ratio) techniques.
4	-	4	B) Metal - ligand equilibria in solution	
<b>Month May</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	-	Theory and Practical Examinations.
-	-	-		

*S. D. Kharade*

Dr. S. D. Kharade



*S. D. Kharade*

Head

Department Of Chemistry

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**Vivekanand College, Kolhapur (Empowered Autonomous)****Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

Name of the teacher: Dr. S. D. Kharade

Programme: M.Sc. II Semester IV

Subject: Chemistry Course Title: Inorganic Chemistry

Month December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	-	Theory and Practical Examination of previous term
-	-	-		
Month January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	<b>Unit I: Inorganic Reaction Mechanism</b>	Types of Mechanisms: Basic concepts as stability and lability, stability constants; HSAB principle, chelate effect, Macrocyclic effect; Ligand transfer and electron transfer reactions in coordination compounds, Intimate and stoichiometric mechanism of ligand substitution. Substitution in square planar complexes: trans effect, trans series, applications of trans effect., Electron Transfer reactions: Potential energy diagrams as a conceptual tool, Marcus equation, Types of and factors affecting electron transfer reactions.
16	4*3 = 12	28		
Month February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	<b>Unit II: Reaction Mechanism of Transition Metal complexes</b>	Substitution reaction, reactions of Transition Metal complexes, kinetics and mechanism of substitution reactions of octahedral complexes, Stereochemical aspects of substitution
16	4*3=12	28		

				reaction of Octahedral Complexes: Stereochemical changes in dissociation (SN2) and displacement (SN2) mechanism through various geometries of coordination compounds. Isomerization and racemization reactions in octahedral complexes. steric effects on substitutions.
<b>Month March</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	<b>UNIT III: Air Pollution and Control Methods</b>	Introduction, Source Correction Methods, Control of Particulate emissions, Selection of a particulate collector, Control of Gaseous Pollutants – NO <sub>x</sub> and Sox, Removal of H <sub>2</sub> S, Control of CO-pollution, Control of Hydrocarbon emission, Control of Pollutant emission from Mobile sources
16	4*3=12	28		
<b>Month April</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	<b>UNIT IV: Water Pollution and Monitoring Control Methods</b>	Introduction A. Sewage and Industrial Waste, COD and BOD Estimation methods, Toxic Heavy metal Analysis – Cd, Hg, As, Pb and Cr Control Methods: Water Softening and Municipal Water Purification B. Techniques in Environmental Analysis ND-IR, FT- IR, AAS, ICT- AES, GCMS, HPLC, Anodic Stripping, Voltametry, etc.
16	4*3=12	28		
<b>Month May</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	-	Theory and Practical Examinations.
-	-	-		

*S. D. Kharade*

Dr. S. D. Kharade



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**Vivekanand College, Kolhapur**  
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**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

Name of the teacher: Miss. S. N. Inamdar

Programme: M.Sc. II Semester IV

Subject: Chemistry Course Title: Inorganic Chemistry

Month: November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Advanced Synthetic Methods of Inorganic Nanomaterials	General Introduction to Nanomaterials, Nanoscience and nanotechnology, History.
2	-	2		
Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Advanced Synthetic Methods of Inorganic Nanomaterials	Chemical bath deposition: ionic and solubility products, preparation of binary semiconductors.
4	-	4		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Advanced Synthetic Methods of Inorganic Nanomaterials	Electrodeposition: Deposition mechanism and preparation of compound thin film.
4	-	4		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Advanced Synthetic Methods of Inorganic Nanomaterials	successive ionic layer adsorption reaction method (SILAR), Co-precipitation, Micelles-microemulsions, Reduction method.
5	-	5		

**Vivekanand College, Kolhapur**  
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Department of Chemistry  
Academic Year: 2023-24


**Annual Teaching Plan**

Name of the teacher: Miss. S. N. Inamdar


Programme: M.Sc. II Semester IV

Subject: Chemistry Course Title: Inorganic Chemistry

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Nanotoxicity and Biosafety	Introduction to Nanotoxicology, Nanoetymology, Nanotoxicology challenges
4	-	4		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Nanotoxicity and Biosafety	Physicochemical characteristic dependent toxicology, Epidemiological evidences.
6	-	6		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Nanotoxicity and Biosafety	Assessment of nanomaterial toxicity: In vitro toxicity assessment-cell viability.
4	-	4		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Nanotoxicity and Biosafety	Mechanism of nanotoxicity, in vivo toxicity assessment.
2	-	2		

  
Miss. S. N. Inamdar



  
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Department of Chemistry  
Academic Year: 2023-24

**Annual Teaching Plan**

Name of the teacher: Miss. S. N. Inamdar

Programme: M.Sc. II Semester IV

Subject: Chemistry Course Title: Analytical Chemistry

Month: November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Analysis of cosmetics products	Introduction to cosmetics, definition, types of cosmetics, background, development in cosmetic industry, issues in cosmetic industries (contamination and adulteration), future scope and role of analytical chemistry.
2	-	2		
Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Analysis of cosmetics products*	A) Analysis of cream and lotions: Composition of creams and lotions, determination of water, propylene glycol, non-volatile matter and ash content; estimation of borates, carbonates, sulphates, phosphates, chlorides, ammonia, nitromethane
4	-	4		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Analysis of cosmetics	oxalic acid, 4- hydroxy benzoic

4	-	4	products-	acid, sodium iodate, freeformaldehyde, H <sub>2</sub> O <sub>2</sub> , titanium and zinc oxides. Practical applications and examples in analytical chemistry and research. B) Analysis of face powder: Composition of face powder, estimation of boric acid, Mg, Ca, Zn, Fe, Al and Ba.
<b>Month: February</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	Analysis of cosmetics products	Analysis of deodorants and antiperspirants-composition, analysis of fats and fatty acids, boric acid, magnesium, calcium, zinc, iron, titanium, aluminium, phenol, methanamine, hexachlorophenone, sulphonates, urea, etc. Practical applications and examples in analytical chemistry and research.
5	-	5		



*S. N. Inamdar*  
Miss. S. N. Inamdar

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**Department of Chemistry**  
Academic Year: 2023-24


**Annual Teaching Plan**

Name of the teacher: Miss. S. N. Inamdar


Programme: M.Sc. I Semester II

Subject: Chemistry Course Title: Inorganic Chemistry (elective) & Physical Chemistry (Practical)

Month: November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Oxygen Transport and Storage	Hemocyanin and hemerythrin
1	8	9		
Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Oxygen Transport and Storage	Synthetic oxygen carriers: Collmans compound; Vaska's complex. Co(II) Schiff base complexes and Perfluorochemicals (PFCs),
4	8	12		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Oxygen Transport and Storage	Perutz mechanism for structural changes in porphyrin ring system, Oxygenation and deoxygenation,
5	12	17		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Oxygen Transport and Storage	Oxygen adsorption isotherm and cooperativity, Role of globin chain in haemoglobin, Siderophores, Vanadium compounds as insulin mimetic agents in the treatment of diabetics.
5	12	17		

  
Miss. S. N. Inamdar



  
Dr. S. D. Shirke  
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**Vivekanand College, Kolhapur (Empowered Autonomous)**

**Annual Teaching Plan**

**Academic Year: 2023-24      Semesters: M.Sc. I Sem-II      Department: Chemistry**

**Subject: Chemistry      Course Title: Analytical Chemistry (Elective)**

**Name of the Teacher: Miss. P. A. Gholap**

Month: December			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Advanced Liquid Chromatographic Techniques	<ul style="list-style-type: none"> <li>• Introduction</li> </ul>
04	-	04		
Month: January			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Advanced Liquid Chromatographic Technique	<ul style="list-style-type: none"> <li>• a) High Performance Liquid Chromatography (HPLC) and Ultra Performance Liquid Chromatography (UPLC)</li> </ul>
04	-	04		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practical	Total	Advanced Liquid Chromatographic Technique	<ul style="list-style-type: none"> <li>• b) Super critical fluid chromatography (SCFC)</li> </ul>
04	-	04		
Month: March			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Advanced Liquid Chromatographic Technique	<ul style="list-style-type: none"> <li>• c) Liquid Chromatography-Mass Spectrometry interface</li> </ul>
03	-	03		
Month: April				<ul style="list-style-type: none"> <li>• Practical and theory Examination</li> </ul>

*P.A. Gholap*

Miss. P. A. Gholap.



*S.D. Shrike*

Dr. Mrs. S. D. Shrike  
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**Vivekanand College, Kolhapur (Empowered Autonomous)**

**Annual Teaching Plan**

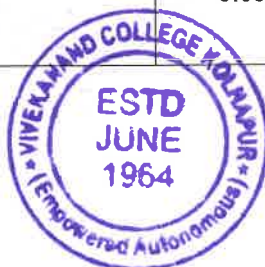
**Academic Year: 2023-24      Semesters: M.Sc. I Sem-II      Department: Chemistry**

**Subject: Chemistry      Course Title: Physical chemistry (Mandatory)**

**Name of the Teacher: Miss. P. A. Gholap**

Month: December			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Thermodynamics	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Practical absolute entropies.</li> <li>• Entropies of phase transition.</li> <li>• Maxwell relations</li> <li>• Thermodynamics of nonelectrolyte solutions.</li> <li>• Raoult's law.</li> <li>• Duhem-Margules equation</li> <li>• Thermodynamic equation of state.</li> </ul>
08	-	08		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Thermodynamics	<ul style="list-style-type: none"> <li>• Thermodynamics of nonelectrolyte solutions.</li> <li>• Raoult's law.</li> <li>• Duhem-Margules equation</li> <li>• Thermodynamic equation of state.</li> <li>• Gibbs-Duhem equation.</li> <li>• Chemical potential</li> <li>• Henry's law</li> <li>• Excess and mixing thermodynamic properties.</li> <li>• Numerical Problems</li> </ul>
07	-	07		
Month: February			Module/Unit:	Sub-units planned
Lectures	Practical	Total	Electrochemistry	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Activity and Activity coefficients</li> <li>• Types of electrodes</li> <li>• Determination of activity coefficients of an electrolyte using concentration cells</li> <li>• Instability constant of silver ammonia complex.</li> </ul>
08	-	08		
Month: March			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Electrochemistry	<ul style="list-style-type: none"> <li>• Acid and alkaline storage batteries,</li> <li>• Abnormal ionic conductance of hydroxyl and hydrogen ions</li> <li>• Electro-osmosis,</li> <li>• Electrophoreses.</li> <li>• Zeta potentials and its determination by electrophoresis.</li> </ul>
07	-	07		

Miss. P. A. Gholap.



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**Vivekanand College, Kolhapur (Empowered Autonomous)**

**Annual Teaching Plan**

**Academic Year: 2023-24**

**Semesters: M.Sc. II, Sem-IV Department: Chemistry**

**Subject: Chemistry**

**Course Title: Analytical Chemistry  
(INDUSTRIAL AND ANALYTICAL CHEMISTRY)**

**Name of the Teacher: Miss. P. A. Gholap**

Month: December			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Analysis of Commercial materials	<ul style="list-style-type: none"> <li>• Introduction</li> <li>• Analysis of explosive materials</li> </ul>
04	-	04		
Month: January			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Analysis of Commercial materials	<ul style="list-style-type: none"> <li>• Analysis of explosive materials, TNT</li> <li>• Analysis of explosive materials, RDX</li> <li>• Analysis of explosive materials, lead azide</li> </ul>
04	-	04		
Month: February			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Analysis of Commercial materials	<ul style="list-style-type: none"> <li>• Analysis of explosive materials EDNA (ethylene dinitramide).</li> <li>• Analysis of conducting polymer, resins and rubber.</li> </ul>
04	-	04		
Month: March			Module/Unit:	Sub-units planned
Lectures	practical's	Total	Analysis of Commercial materials	<ul style="list-style-type: none"> <li>• Analysis of luminescent paints,</li> <li>• Analysis of lubricants and adhesive.</li> </ul>
03	-	03		
Month: April				• Practical and theory Examination



Miss. P. A. Gholap




Dr. Mrs. S. D. Shirke

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Vivekanand College, Kolhapur (Empowered Autonomous)

Annual Teaching Plan

Academic Year: 2023-24

Semesters: M.Sc. II, Sem-IV

Department: Chemistry

Subject: Chemistry

Course Title: Inorganic Chemistry

Name of the Teacher: Miss. P. A. Gholap

Month: December			Module/Unit:	Sub-units planned
Lectures	Practical's	Total	X-ray Diffraction Techniques	<ul style="list-style-type: none"> <li>Introduction</li> <li>A] X-ray powder diffraction (XRD)</li> </ul>
04	-	04		
Month: January			Module/Unit:	Sub-units planned
Lectures	Practical's	Total	X-ray Diffraction Techniques	<ul style="list-style-type: none"> <li>Diffraction of X-rays with powder diffraction, Instrumentation.</li> <li>Identification of compounds using powder diffraction.</li> </ul>
04	-	04		
Month: February			Module/Unit:	Sub-units planned
Lectures	practical's	Total	X-ray Diffraction Techniques	<ul style="list-style-type: none"> <li>Determination of parameters using XRD</li> <li>Determination of cubic crystal structure.</li> </ul>
04	-	04		
Month: March			Module/Unit:	Sub-units planned
Lectures	practical's	Total	X-ray Diffraction Techniques	<ul style="list-style-type: none"> <li>Single Crystal X-ray Diffraction:</li> <li>X-ray crystal structures in the literature.</li> </ul>
03	-	03		
Month: April				<ul style="list-style-type: none"> <li>Practical and theory Examination</li> </ul>

Miss. P. A. Gholap



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**Annual Teaching Plan**

**Academic Year: 2023-24      Semesters: M.Sc. I Sem-II      Department: Chemistry**

**Subject: Chemistry      Course Title: Analytical Chemistry Practical's**

**Name of the Teacher: Miss. P. A. Gholap**

Month: September			Module/Unit:	Sub-units planned
Lectures	Practical's	Total	-	1.To verify the Beers- Lamberts law and determine the concentration of given solution using colorimetry. 2. To determine percentage purity of given olefinic compound by bromination method. 3.To estimate percentage of copper from supplied sample of copper fungicide.
-	12	12		
Month: October			Module/Unit:	Sub-units planned
Lectures	Practical's	Total	-	4. To estimate amount of calcium present in given sample of plaster of Paris. 5. To determine solubility of calcium oxalate in presence of different concentration of HCl. 6. To determine standard deviation from redox titration of Fe <sup>2+</sup> i.e. is iron solution against standard potassium dichromate.
-	12	12		
Month: November			Module/Unit:	Sub-units planned
Lectures	Practical's	Total	-	7. To determine the Hardness of given water sample. 8. To determine the chlorine content available in given bleaching powder.
-	08	08		

*P.A. Gholap*

Miss. P. A. Gholap



*S.D. Shirke*

Dr. Mrs S. D. Shirke  
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**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. II Semester IV

**Subject:** Chemistry

**Course Title:** Analytical Chemistry

<b>Month August</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Ion Chromatography	Principles, structure and characteristics of resins, eluent, suppressor columns and detectors used in Ion Chromatography, Practical applications and examples in analytical chemistry and research.
4	-	4		
<b>Month September</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Ion Chromatography	Commercial scope, analytical applications, environmental speciation by Ion Chromatography.
4	-	4		
<b>Month October</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Ion Chromatography	Ion Chromatography, Practical applications and examples in analytical chemistry and research.
4	-	4		
<b>Month November</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>

Lectures	Practicals	Total	UNIT-III: Ion Chromatography	Ion Chromatography, Practical applications and examples in analytical chemistry and research..
4	-	4		



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Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Ion Chromatography, Practical applications and examples in analytical chemistry and research..
4	-	4		



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**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. II Semester IV

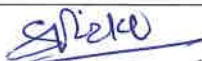
**Subject:** Chemistry

**Course Title:** Analytical Chemistry

Month August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	X-ray spectroscopy	Introduction, X-Ray generation, Properties of X-radiation
4	-	4		
Month September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	X-ray spectroscopy	X-Ray, Instrumentation, X-Ray Absorption,
4	-	4		
Month October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	X-ray spectroscopy	Fluorescence and Diffraction methods of analysis and their applications.
4	-	4		
Month November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	X-ray spectroscopy	Fluorescence and Diffraction methods of analysis and their applications.
4	-	4		

  
MS. V. S. Rajmane



  
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**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. II Semester IV

**Subject:** Chemistry

**Course Title:** Organic Chemistry

Month August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT – IV A) Kinetic and thermodynamic control of reactions	Nitration and Sulphonation of naphthalene, Wittig, Enolization,
4	-	4		
Month September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT – IV A) Kinetic and thermodynamic control of reactions	Friedel-Crafts and Diels Alder reactions.
4	-	4		
Month October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Non-classical carbocations:	Non-classical carbocations
4	-	4		
Month November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Non-classical carbocations:	Formation, stability and reactivity.
4	-	4		

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Vivekanand College, Kolhapur (Autonomous)

Department of Chemistry

Academic Year: 2023-24

Annual Teaching Plan

Name of the Teacher: MS. V. S. Rajmane.

Programme: M.Sc. II Semester III

Subject: Chemistry

Course Title: Analytical Chemistry

Month July			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit IV: A) Ion selective electrodes	Terminology, types and construction of electrodes, glass electrode
3	-	3		
Month August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	A) Ion selective electrodes	solid state and precipitate electrodes, liquid – liquid membrane electrodes
12	4*3 = 12	24		
Month September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	A) Ion selective electrodes	Enzyme and gas electrodes, and applications. Organic synthesis relating to nucleophilic and electrophilic attack on ligands.
12	4*3=12	24		
Month October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit IV: Electrophoresis	Introduction: paper electrophoresis: Technique, factors
12	4*3=12	28		

				affecting migration of ions,
<b>Month November</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
			Unit IV: Electrophoresis	Capillary and zone electrophoresis and applications.
12	4*3=12	24		
<b>Month December</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	Unit IV: Electrophoresis	Practical applications in analytical chemistry and research.
12	-	12		



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Vivekanand College, Kolhapur (Autonomous)

Department of Chemistry

Academic Year: 2023-24

Annual Teaching Plan

Name of the Teacher: MS. V. S. Rajmane.

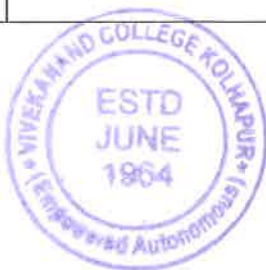
Programme: M.Sc. II Semester III


Subject: Chemistry

Course Title: Organic Chemistry

Month August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit IV: (A) Five- Study of five and six-membered heterocycles containing one membered heterocycles	Synthesis and reactions of (i) Pyrrol (ii) Furan, (iii) Thiophene, (iv) Benzopyrroles
4	-	4		
Month September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit IV: (A) Five- Study of five and six-membered heterocycles containing one membered heterocycles	(v) Benzofurans, and (vi) Benzothiophenes.
4	-	4		
Month October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	(B)Six-membered heterocycles with one heteroatom	Synthesis and reactions of (i) Pyridine,
4	-	4		
Month November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	(B)Six-membered heterocycles with one heteroatom	(ii)Quinoline, and (iii) Coumarin.
4	-	4		

  
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**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. I Semester I

**Subject:** Chemistry

**Course Title:** Analytical Chemistry

<b>Month August</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT III: Electroanalytical Techniques	Polarography: Introduction, Instrumentation, Ilkovic equation and its verification. Polarographic measurements, Dropping mercury electrode
4	-	4		
<b>Month September</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT III: Electroanalytical Techniques	Determination of half wave potential, qualitative and quantitative applications
4	-	4		
<b>Month October</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT III: Electroanalytical Techniques	Amperometry: Basic principles, instrumentation, Amperometric titration curves, Amperometric indicators procedure for Amperometric titrations, Evaluation of amperometry in research and analytical applications.
4	-	4		

Month November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT III: Electroanalytical Techniques	Voltammetry: Voltammetric methods of analysis, basic principles, instrumentation, voltammetric measurements, Volta metric techniques, current in voltammetry, shape of voltammograms, quantitative and qualitative aspects of voltammetry, quantitative applications, characterization applications, Evaluation of CV in research and analytical applications
4	-	4		



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**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. II Semester I

**Subject:** Chemistry

**Course Title:** Analytical Chemistry

<b>Month August</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	Unit IV: Atomic Absorption & Inductively Coupled Plasma Spectroscopy -	Introduction, Principal, the difference between AAS and FES, Advantages of AAS over FES, advantages and disadvantages of AAS
4	-	4		
<b>Month September</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	Unit IV: Atomic Absorption & Inductively Coupled Plasma Spectroscopy	Instrumentation, Single and double beam AAS, detection limit and sensitivity, Interferences, applications. Graphite furnace atomic absorption spectroscopy, general description
4	-	4		
<b>Month October</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	Unit IV: Atomic Absorption & Inductively Coupled Plasma Spectroscopy	Advantages and disadvantages. Flame photometry, Cold Vapor Mercury, Hydride Generation, Spark emission, challenges and limitations.
4	-	4		

Month November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	Unit IV: Inductively Coupled Plasma Spectroscopy:	Introduction, Nebulization Torch, Plasma, Instrumentation, Interferences, and Applications.  Problems: Simple problems based on AAS and ICP
4	-	4		



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**Vivekanand College, Kolhapur (Autonomous)**

Department of Chemistry

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. I Semester II

**Subject:** Chemistry

**Course Title:** Analytical Chemistry

Month August			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-I: Study of following chromatographic Techniques	Partition chromatography: general principle, classification of chromatographic Techniques nature of partition forces, chromatographic behaviors of solutes, columns efficiency and resolution.
4	-	4		
Month September			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-I: Study of following chromatographic Techniques	Thin layer chromatography Thin Layer Chromatography general principle, and Applications.
4	-	4		
Month October			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-I: Study of following chromatographic Techniques	Classification of chromatographic Techniques, Thin Layer Chromatography Techniques.
4	-	4		

Month November			Module/Unit:	Sub-units planned
Lectures	Practicals	Total	UNIT-I: Study of following chromatographic Techniques	Column chromatograph: principle and theory, adsorption and partition methods, advantages, Disadvantages and application of Column chromatograph
4	-	4		



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(EMPOWERED AUTONOMOUS)

**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. I Semester II

**Subject:** Chemistry

**Course Title:** Analytical Chemistry

<b>Month August</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Principles, structure and characteristics of resins, eluent, suppressor columns and detectors used in Ion Chromatography, Practical applications and examples in analytical chemistry and research.
4	-	4		
<b>Month September</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Commercial scope, analytical applications, environmental speciation by Ion Chromatography.
4	-	4		
<b>Month October</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Ion Chromatography, Practical applications and examples in analytical chemistry and research.
4	-	4		
<b>Month November</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>

**Vivekanand College, Kolhapur (Autonomous)**

**Department of Chemistry**

Academic Year: 2023-24

**Annual Teaching Plan**

**Name of the Teacher:** MS. V. S. Rajmane.

**Programme:** M.Sc. I Semester II

**Subject:** Chemistry

**Course Title:** Analytical Chemistry

<b>Month August</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Principles, structure and characteristics of resins, eluent, suppressor columns and detectors used in Ion Chromatography, Practical applications and examples in analytical chemistry and research.
4	-	4		
<b>Month September</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Commercial scope, analytical applications, environmental speciation by Ion Chromatography.
4	-	4		
<b>Month October</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>
Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Ion Chromatography, Practical applications and examples in analytical chemistry and research.
4	-	4		
<b>Month November</b>			<b>Module/Unit:</b>	<b>Sub-units planned</b>



Lectures	Practicals	Total	UNIT-III: Chromatographic Techniques.	Ion Chromatography, Practical applications and examples in analytical chemistry and research..
4	-	4		



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*SPD*

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